

China factory High Purtiy Hcds 99.9% Si2cl6 Hexachlorodisilane

Basic Information

. Place of Origin: China . Brand Name: CMC COA · Certification: Si2cl6 Model Number: • Minimum Order Quantity: 1kg • Price: US \$40/kg · Packaging Details: Cylinder/Tank • Delivery Time: 15 days Payment Terms: L/C, T/T . Supply Ability: 50000kg/month



Product Specification

Product Name: Hexachlorodisilane

Model No.: Si2cl6 • Transport: By Sea 99.9% • Purity: • Transport Package: Cylinder Specification: 40L, 200L • Trademark: CMC • Origin: China • HS Code: 2812190091 • Supply Ability: 100t/Year 7783-82-6 CAS No.: Si2cl6 • Formula: . EINECS: 7783-82-6



More Images

Constituent:

• Grade Standard:





Industrial Pure Air

Industrial Grade





Product Description

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Hexachlorodisilane (Si2Cl6) is an inorganic compound consisting of two silicon (Si) atoms bonded together with six chlorine (Cl) atoms. It is a colorless gas that is primarily used as a precursor in the production of silicon-based materials. Here are some key points about hexachlorodisilane

Structure and Properties: Hexachlorodisilane has a tetrahedral molecular structure, with each silicon atom bonded to three chlorine atoms and the other silicon atom. It is a volatile and reactive compound, and its boiling point is around $140\,^{\circ}\text{C}.$

Synthesis and Production: Hexachlorodisilane is typically synthesized by the reaction of silicon tetrachloride (SiCl4) with elemental silicon (Si) or silicon powder at high temperatures. It can also be produced through the reaction of silicon with chlorine gas (Cl2).

Applications:

Semiconductor Industry: Hexachlorodisilane is a crucial precursor in the production of various silicon-based thin films for the semiconductor industry. It is used in chemical vapor deposition (CVD) and atomic layer deposition (ALD) processes to deposit silicon-containing layers with high purity and uniformity.

Silicon Nitride Production: Hexachlorodisilane can be used as a precursor for the synthesis of silicon nitride (Si3N4) ceramics. It reacts with ammonia (NH3) to form silicon nitride, which is used in the manufacturing of ceramics, cutting tools, and protective coatings.

Surface Modification: Hexachlorodisilane is employed as a surface treatment agent to modify the surface properties of materials. It can enhance $hydrophobic\ or\ hydrophilic\ characteristics,\ improve\ adhesion,\ or\ provide\ chemical\ resistance\ to\ surfaces.$

Safety Considerations: Hexachlorodisilane is a highly reactive and volatile compound. It is corrosive to metals and can cause severe burns upon contact with the skin or eyes. It is also toxic if inhaled or ingested. Proper handling, storage, and personal protective equipment should be used when working with hexachlorodisilane or any other hazardous chemicals.

Hexachlorodisilane gas is an important precursor in the semiconductor industry and the production of silicon-based materials. Its reactivity and ability to deposit high-quality silicon films make it valuable for various technological applications. However, it should be handled with caution due to its hazardous properties.

Basic Info.

Model NO.	Si2Cl6	Grade Standard	Electron Grade
Transport Package	Cylinder, Canister	Specification	40L, 200L
Trademark	СМС	Origin	Suzhou, China
HS Code	2812190091	Production Capacity	100t/Year

Specifications:

IUPAC name Hexachlorodisilane Other names Disilicon hexachloride

Identifiers

CAS No.: 13465-77-5 EC No.: 236-704-1

Properties

Molecular Formula: Si2Cl6 Molar mass: 268.88 g/mol Appearance: Colorless liquid

Melting point: ≤20 °C

144 °C (291 °F; 417 K) **Boiling point:**

Flash point: >93°C

Vopor density(Air=1): Relative density(Water=1): 1.562

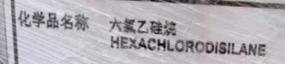
Sample Test:

Test Items	Units	Specifications	Test Result
Assay by GC	wt%	≥99.9	99.905
Li	ng/g	≤0.5	< 0.05
Na	ng/g	յ ≤0.5	< 0.05
Mg	ng/g	j ≤0.5	< 0.05
Al	ng/g	j≤1.0	0.35
K	ng/g	յ ≤0.5	0.08
Ca	ng/g	j ≤0.5	0.16
Ti	ng/g	j≤1.0	0.18
Cr	ng/g	յ ≤0.5	< 0.05
Mn	ng/g	j ≤0.5	< 0.05
Fe	ng/g	j≤ 1.0	0.48
Co	ng/g	j ≤0.5	< 0.05

Detailed Photos







CAS No.: 13496-77-6 含量。100%





造成严重皮肤灼伤和眼损伤。

【預防推施】

- ·不要吸入粉尘/烟/气体/烟雾/蒸气/喷雾。
- 作业后彻底清洗。
- · 戴防护手套/穿防护服/戴防护跟罩/戴防护面具。

【事故响应】

- · 立即呼叫中毒急救中心/医生。
- 沾染的衣服清洗后方可重新使用。
- •如读吸入,将受人转移到空气新鲜处,保持
- 知误吞噬,漱口,不要诱导呕吐。
- ·如皮肤(或头发) 結集,立即去除/脱掉所有結集的衣服。用水清洗皮肤或淋形。 、如迎入眼睛。用水小心冲洗几分钟。如戴隐形眼镜照可方便地取出,取出建形膜鏡。 建核冲 贵

【安全领存】

• 存放处项加锁

【皮井处里】

按照地方/区域/国家/国际规章处置内装物/容器。

请参阅化学品安全技术说明书

铁边翼。落刻中坡高到技有观台词一分会司

ALIE: 0379-68606102

地 並、 務期市市及科技提区社外大道政路 1 号 M M. 471000

化學事故國內股惠皆被电话。0379-69608119





Shanghai Kemike Chemical Co.,Ltd

+86 18762990415

williamchen@cmc-chemical.com

@ gascylindertank.com